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with sodium bicarbonate or calcium hydroxide, absorb with inert dry material, sweep or vacuum up and place in a suitable container for proper disposal. Wash spill area with soap and water. Avoid runoff into storm sewers and ditches which lead to waterways.

Handling & Storage CORROSIVE STORAGE CODE WHITE

Read label on container before using. Do not wear contact lenses when working with chemicals. Keep container tightly closed. For laboratory use only. Not for drug, food or household use. Keep out of reach of children. Handling: Use with adequate ventilation. Avoid contact with eyes, skin and clothing. Avoid ingestion. Do not inhale vapors, spray or mist. Wash thoroughly after handling. Remove and wash clothing before reuse. Storage: Store in a cool, dry, well-ventilated area away from incompatible substances. Protect from physical damage and sunlight. Protect from moisture.

Section 8 Exposure Controls / Personal Protection

Engineering controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower and fire extinguishing material. Personnel should wear safety glasses, goggles, or faceshield, lab coat or apron, appropriate protective gloves. Use adequate ventilation to keep airborne concentrations low.

Respiratory protection: Use a chemical fume hood and/or wear a NIOSH/MSHA-approved respirator.

Section 9 Physical & Chemical Prope	rties.
Physical state: Liquid.	Boiling point: ~100°C (~212°F) [water]
Appearance: Clear, colorless.	Freezing / Melting point: ~0°C (~32°F) [water]
Odor: Pungent odor.	Decomposition temperature: N/A
pH: N/A	Solubility: Soluble
Vapor pressure (mm Hg): 14 [water]	Specific gravity (H <sub>2</sub> O = 1): 1.0 [water]
Vapor Density (Air = 1); 0.7 [water]	Percent volatile (%): 100%
Evaporation rate ( = 1); N/A	Molecular formula: Mixture.
Viscosity: N/A	Molecular weight: Mixture.
Section 10 Stability & Reactivity	
Chemical stability: Stable	Hazardous polymerization: Will not occur.
Conditions to avoid: Containers may burst when hea	ited. Avoid contact with water.
Incompatibilities with other materials: Metals, base hydroxides, amines, carbonates, cyanides, sulfides, su	
Hazardous decomposition products: Hydrogen, chlo	orine.
	4. ( <sup>**</sup>
Section 11 Toxicological Information	
	drochloric acid can cause immediate pain and burns of the
mouth, throat, esophagus and gastrointestinal tract. Va	apors are irritating and may cause damage to the eyes.
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mouth; throat, esophagus and gastrointestinal tract. Vapors are irritating and may cause damage to the eyes. Splashes may cause severe burns and permanent eye damage. Can cause redness, pain, and severe skin burns. Inhalation of vapors can cause coughing, choking, inflammation of the nose, throat, and upper respiratory tract.

RTECS #: MW4025000 [Hydrochloric acid]

## Section 12 Ecological Information The methods for determining the biological degradability are not applicable to inorganic substances. Harmful ecological effects due to the pH shift are expected.

Section 13. Disposal Considerations These disposal guidelines are intended for the disposal of catalog-size quantities only. Federal regulations may apply to empty container. State and/or local regulations may be different. Dispose of in accordance with all local, stat and federal regulations or contract with a licensed chemical disposal agency.

Section 14 Transport Information
UN/NA number: UN1789
Shipping name: Hydrochloric acid
Hazard class: 8
Packing group: 111
Exceptions: Ltd Qty ≤ 5 Lt.
Section 15 Regulatory Information
TSCA-listed, EINECS-listed (231-595-7), RCRA code D002.
Section 16 Additional Information

The information contained bacain is furnished without warranty of any kind. Employers should use this information only as a supplement